

Water And Electrolytes Balance

Dr.Komal Chavda

Dep. of Physiology

Batch 2021

Body Water

- ▶ Water is the **chief constituent of human body.**
- ▶ Water is the **chief solvent of body.**
- ▶ Water **comprises 60-70%** of total body weight
- ▶ **Human body cannot exist without Water.**

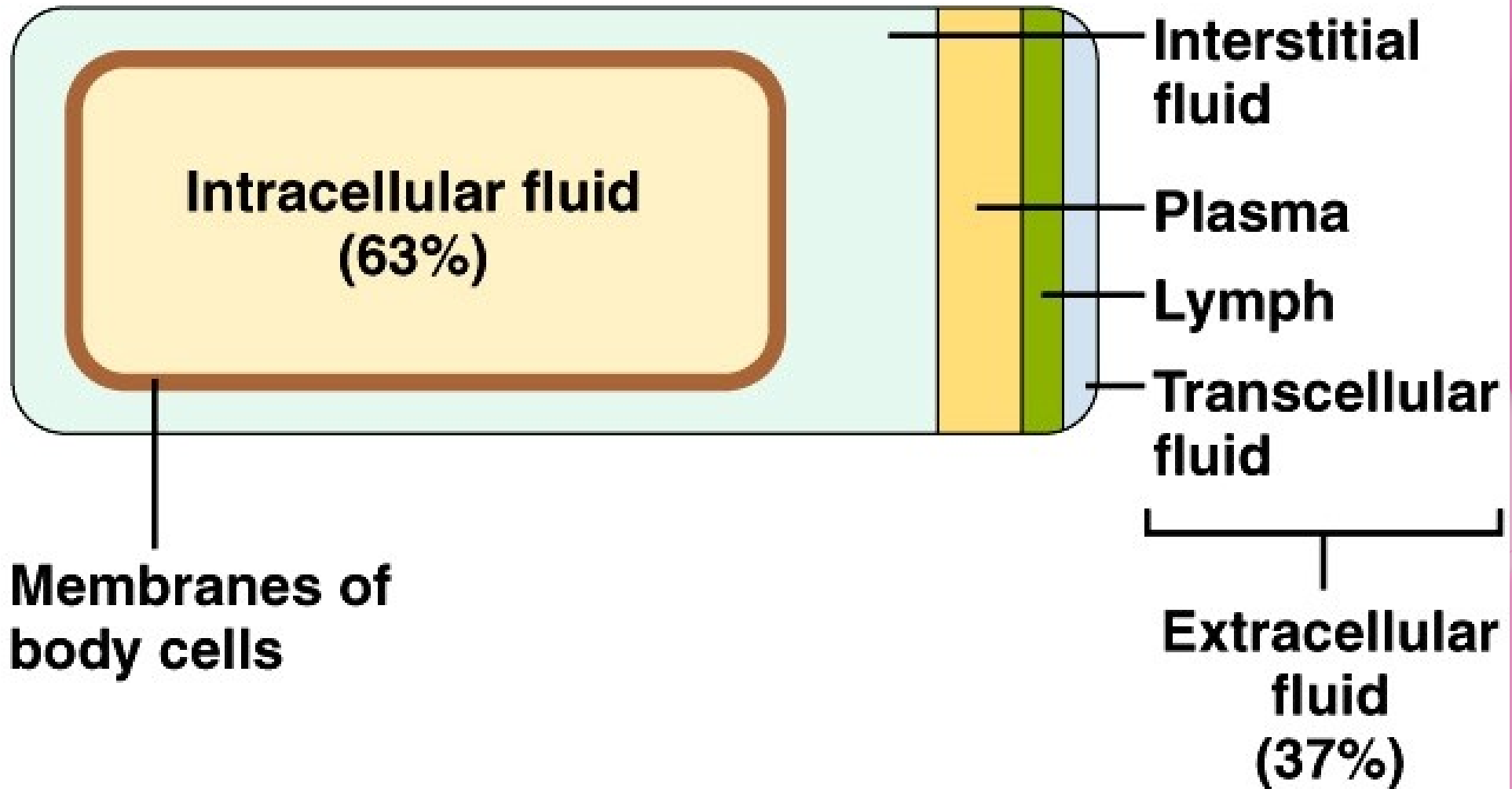
Sources Of Body Water

- ▶ Drinking Water, Beverages -1000 to1500 ml
- ▶ Water from Cooked Foods
- ▶ Water intake through mouth is highly variable 1-5 Liters this depend on :
 - Social habits
 - Climatic condition

Endogenous Sources of Water

- ▶ **Metabolic Water - 400 ml**
- ▶ Produced during metabolism oxidation of food substances.
- ▶ In an adult of 70 kg body
- ▶ **Total Body Water – 60 to 70%**
- ▶ **Intracellular Fluid -65 %**
- ▶ **Extracellular Fluid -35%**
 - ▶ Interstitial Tissue Fluid -25%
 - ▶ Plasma /Intra Vascular Fluid -8%
 - ▶ Transcellular Fluid- 2%

Total body water



► Body water content in percentage of a body weight is lowest in.

(A) Well built man

(B) Fat woman

(C) Well nourished child

(D) Fat Man

Functions Of Body Water

- ▶ **Involved in Biochemical reactions**
 - ▶ Water act as reactant in many hydration Hydrolytic reactions of metabolic pathways.
- ▶ **Transporting media of body:**
 - ▶ Transportation of nutrients and waste metabolites through media of blood and tissue floods.
- ▶ **Regulates body temperature**

- ▶ Water transports **Hormones**, Enzymes, **blood platelets**, and **red and white blood cells**
- ▶ Water act as a solvent for **Electrolytes** and **Non electrolytes**
- ▶ Water Facilitates **Digestion** and promoting **Elimination of ingested food**
- ▶ Water serve as a tissue **Lubricant**

Body Water Input and Output

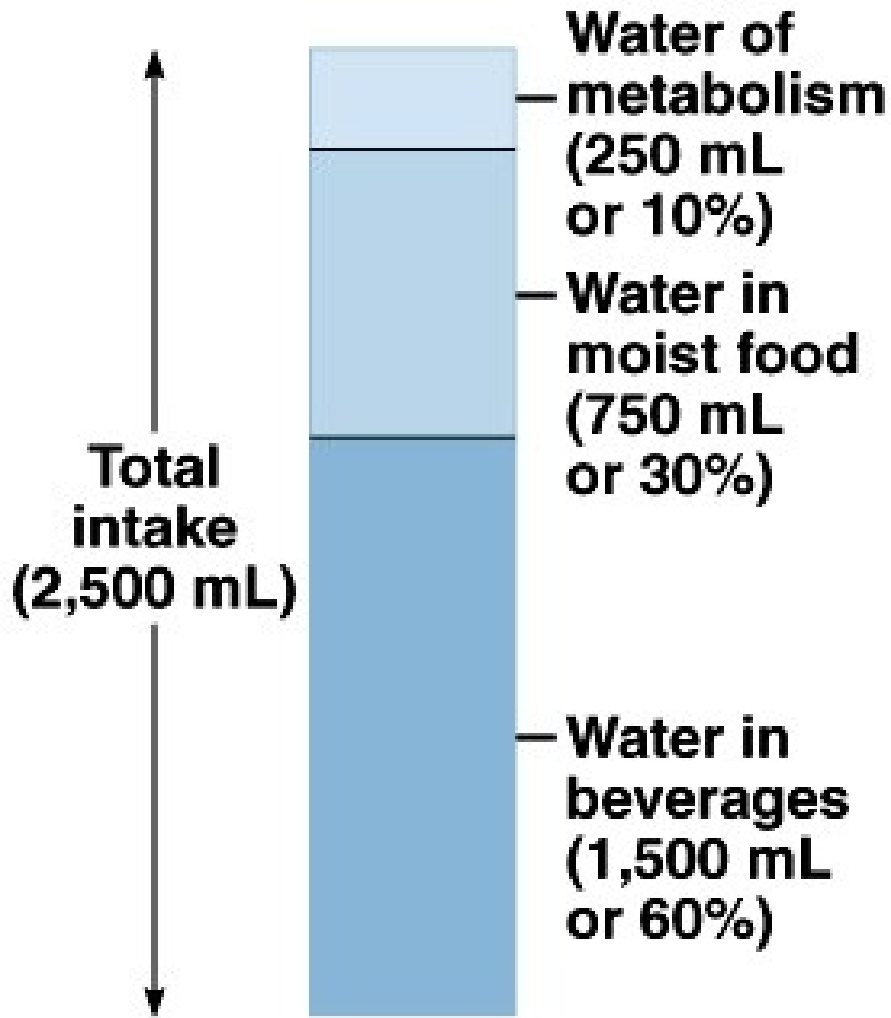
□ Body Water Input

- ▶ Body can gain water by :
 - ▶ **Ingestion** of liquids and moist foods (**2300mL/day**)
 - ▶ **Metabolic** synthesis of water during cellular respiration (**200mL/day**)

Body Water Output

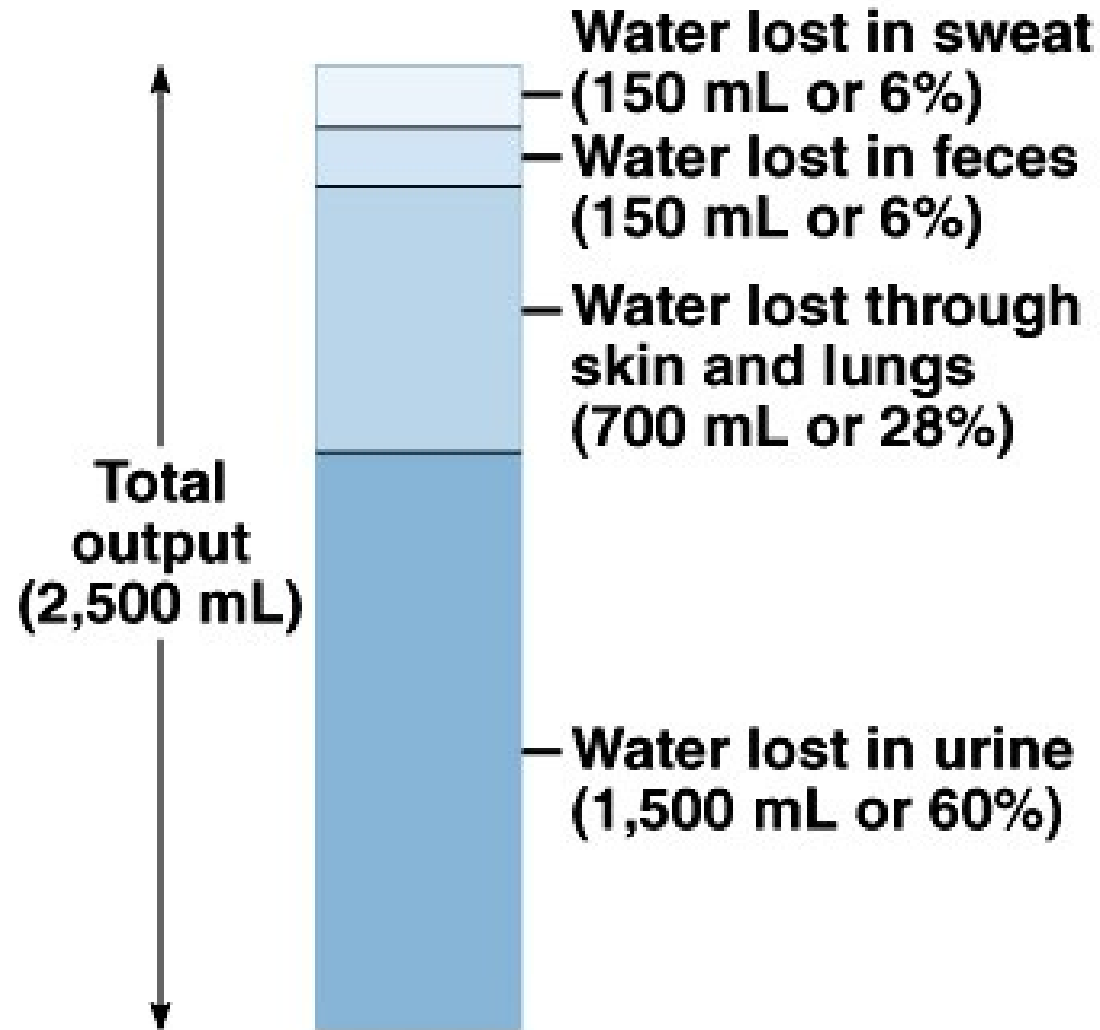
- ▶ Body losses water through:
 - ▶ Kidneys (1500mL/day)
 - ▶ Evaporation from Skin (600mL/day)
 - ▶ Exhalation from Lungs (300mL/day)
 - ▶ Feces (100mL/day)

Average daily intake of water



(a)

Average daily output of water



(b)

BODY ELECTROLYTES

- ▶ Substance when dissolved in solution **dissociates into ions.**
- ▶ These ions are able to **carry an electrical current.**
- ▶ **An Electrolyte** is a substance which **develops an electrical charge when dissolved in water.**
- ▶ Salts like **NaCl** and **KCl** in aqueous solutions gets dissociated to **Charged ions Na^+ and Cl^-** called as **Electrolytes.**
- ▶ The **concentration of these Electrolytes is expressed** as **mEq/L.**

Types Of Electrolytes

- ▶ **CATION** - Positively charged Electrolyte
- ▶ **ANION** - Negatively charged Electrolyte
- ▶ **Predominant Cations and Anions of ECF: Na^+ and Cl^- respectively.**

Distribution Of Body Electrolytes In ECF and ICF

INTRACELLULAR Electrolytes	EXTRACELLULAR Electrolytes
POTASSIUM	SODIUM
MAGNESIUM	CHLORIDE
PHOSPHOROUS	BICARBONATE

Functions Of Body Electrolytes

- ▶ **Electrolytes are well distributed in the body compartments.**
- ▶ Electrolytes in the medium/compartments **produce osmotic pressure.**
- ▶ This osmotic pressure **helps in maintaining water balance.**

ELECTROLYTES

- ▶ **Na⁺**: Most abundant electrolyte in the ECF.
- ▶ **K⁺**: Essential for normal membrane excitability for nerve impulse
- ▶ **Cl⁻**: Regulates osmotic pressure and assists in regulating acid-base balance
- ▶ **Ca²⁺**: Promotes nerve impulse and muscle contraction/relaxation
- ▶ **Mg²⁺**: Plays role in carbohydrate and protein metabolism, storage and use of intracellular energy and neural transmission. Important in the functioning of the heart, nerves, and muscles.

THANK YOU